

12-{[4-(2-Fluorophenyl)piperazin-1-yl]-methyl}-9 α -hydroxy-4,8-dimethyl-3,14-dioxatricyclo[9.3.0.0^{2,4}]tetradec-7-en-13-one

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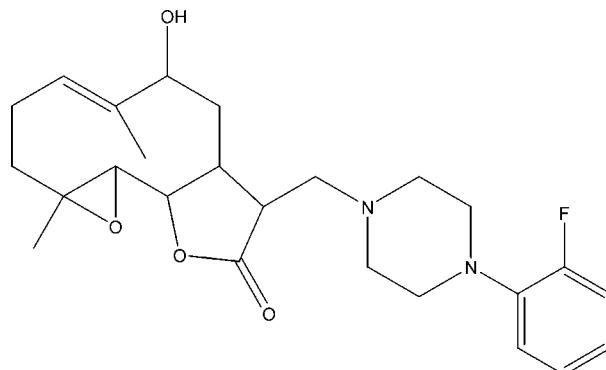
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Key indicators: single-crystal X-ray study; $T = 180\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$; R factor = 0.068; wR factor = 0.197; data-to-parameter ratio = 8.7.

The title compound, $C_{25}H_{33}FN_2O_4$, was synthesized from 9 α -hydroxyparthenolide (9 α -hydroxy-4,8-dimethyl-12-methylene-3,14-dioxatricyclo[9.3.0.0^{2,4}]tetradec-7-en-13-one), which was isolated from the chloroform extract of the aerial parts of *Anvillea radiata*. The asymmetric unit contains two independent molecules. In each molecule, the ten-membered ring displays an approximative chair-chair conformation. Each of the piperazine rings adopts a perfect chair conformation, while both lactone rings show an envelope conformation, one with the C atom bearing the piperazin-1-ylmethyl group as the flap, the other with the junction C atom not attached to the ring O atom as the flap. The dihedral angles between the least-squares planes through the ten- and five-membered rings in the two molecules are similar [19.1 (3) and 16.2 (3) $^\circ$]. An intramolecular O—H \cdots N hydrogen bond stabilizes the molecular conformation. The crystal packing is stabilized by C—H \cdots O hydrogen bonds.

Related literature

For background to the medicinal uses of the plant *Anvillea radiata*, see: El Hassany *et al.* (2004); Qureshi *et al.* (1990). For the reactivity of this sesquiterpene, see: Castaneda-Acosta *et al.* (1997); Hwang *et al.* (2006); Neukirch *et al.* (2003); Neelakantan *et al.* (2009). For ring puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$C_{25}H_{33}FN_2O_4$	$V = 2277.7(8)\text{ \AA}^3$
$M_r = 444.53$	$Z = 4$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
$a = 14.583(2)\text{ \AA}$	$\mu = 0.09\text{ mm}^{-1}$
$b = 6.3366(17)\text{ \AA}$	$T = 180\text{ K}$
$c = 24.697(3)\text{ \AA}$	$0.56 \times 0.36 \times 0.1\text{ mm}$
$\beta = 93.598(14)^\circ$	

Data collection

Agilent Xcalibur Eos Gemini ultra diffractometer	24656 measured reflections
Absorption correction: multi-scan (<i>CrysAlis PRO</i> ; Agilent, 2011)	5050 independent reflections
$T_{\min} = 0.789$, $T_{\max} = 1.000$	4582 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.066$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.068$	1 restraint
$wR(F^2) = 0.197$	H-atom parameters constrained
$S = 1.12$	$\Delta\rho_{\text{max}} = 0.43\text{ e \AA}^{-3}$
5050 reflections	$\Delta\rho_{\text{min}} = -0.37\text{ e \AA}^{-3}$
583 parameters	

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
O1—H1B \cdots N1	0.82	2.19	2.979 (6)	163
O1A—H1A1 \cdots N2A	0.82	2.08	2.882 (6)	164
C1—H1 \cdots O3 ⁱ	0.98	2.47	3.035 (7)	116
C13—H131 \cdots O3 ⁱⁱ	0.97	2.51	3.394 (7)	152
C10—H10 \cdots O1	0.98	2.36	2.861 (6)	111
C10A—H10A \cdots O1A	0.98	2.36	2.860 (6)	111
C13A—H13A \cdots O3A ⁱⁱⁱ	0.97	2.57	3.391 (6)	143
C11A—H11A \cdots O3A ^{iv}	0.98	2.60	3.338 (6)	132
C15—H15F \cdots O1 ^v	0.96	2.42	3.375 (7)	171

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + 1$; (ii) $-x + 1, y - \frac{1}{2}, -z + 1$; (iii) $-x + 2, y - \frac{1}{2}, -z$; (iv) $-x + 2, y + \frac{1}{2}, -z$; (v) $-x + 2, y + \frac{1}{2}, -z + 1$.

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5817).

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supplementary materials

Acta Cryst. (2012). E68, o741–o742 [doi:10.1107/S1600536812006411]

12-{[4-(2-Fluorophenyl)piperazin-1-yl]methyl}-9 α -hydroxy-4,8-dimethyl-3,14-dioxatricyclo[9.3.0.0^{2,4}]tetradec-7-en-13-one

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Comment

Our work lies within the framework of the evaluation of medicinal plants and in particular, *Anvillea radiata*. The main constituent of the chloroform extract of aerial parts of this plant is 9 α -hydroxypartenolide(El Hassany *et al.*, 2004). The reactivity of this sesquiterpene lactone and its derivatives has been the subject of several studies(Castaneda-Acosta *et al.*, 1997; Neukirch *et al.*, 2003; Hwang *et al.*, 2006 Neelakantan *et al.*, 2009), in order to prepare products with high value which can be used in the pharmacological industry. In this context, we have treated the 9 α -hydroxypartenolide with an equivalent amount of 1-(2- fluorophenyl)piperazine and isolated the 9 α -Hydroxy-4,8- dimethyl-12-[(4-(2-fluorophenyl)-piperazin-1-yl)methyl]- 3,14-dioxatricyclo-[9.3.0.0^{2,4}]tetradec-7-en-13-one with a yield of 95%. The structure of this new product was confirmed by its single-crystal X-ray structure. The asymmetric unit contains two crystallographically independent molecules(Fig.1). Each molecule is built up from two fused five-and ten-membered rings with the fluoro-phenylpiperazine group as a substituent. The ten-membered ring displays an approximate chair-chair conformation. Whereas the lactone ring (O2A, C1A…C12A) adopt an envelope conformation, as indicated by the puckering parameters Q = 0.229 (5) Å and φ = 78.9 (11) $^{\circ}$ (Cremer & Pople, 1975), the other lactone ring (O2,C1…C12)shows a twisted conformation with Q = 0.204 (5) Å and φ = 47.9 (14) $^{\circ}$. The piperazine ring, in the two molecules, has a perfect chair conformation with QT = 0.594 (5) Å, θ = 176.5 (5) $^{\circ}$ and φ_2 = 9.0 (9) $^{\circ}$ for the ring (N1A,C17A…C18A)and QT = 0.587 (5) Å, θ = 176.2 (5) $^{\circ}$ and φ = 187.0 (7) $^{\circ}$ for the other piperazine ring (N1,C16…C19).In the first molecule (C1 to C25), the dihedral angle between the mean planes of the ten-membered ring and the lactone ring is 16.2 (3).The corresponding value in the second molecule (C1A to C25A) is 19.3 (3) $^{\circ}$. In the crystal, C—H…O hydrogen bonding links the molecules into sheets lying parallel to the *c* axis (Table 1, Fig.2). In addition, the molecular conformation is stabilized by an O—H…N hydrogen bond between the hydroxy group and a piperazine N atom.

Experimental

The mixture of 9 α -hydroxypartenolide (0.5 g, 2 mmol) and one equivalent of 1-(2- fluorophenyl) piperazine in EtOH (20 ml) was stirred for one night at room temperature. The next day, the reaction was stopped by adding water (10 ml) and extracted three times with ethyl acetate (3 x 20 ml). The combined organic layers were dried over anhydrous MgSO₄, filtered and concentrated under vacuum to give 1 g (1.9 mmol)of the title compound (yield: 95%) which was recrystallized in ethyle acetate.

Refinement

All H atoms were fixed geometrically and treated as riding with O—H = 0.82 Å, C—H = 0.96 Å (methyl), 0.97 Å (methylene), 0.98 Å (methine) with $U_{\text{iso}}(\text{H})$ = 1.2 U_{eq} (methylene, methine) or $U_{\text{iso}}(\text{H})$ = 1.5 U_{eq} (methyl, OH). In the absence of significant anomalous scattering, the absolute configuration could not be reliably determined and thus Friedel pairs

were merged and any references to the Flack parameter were removed.

Computing details

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO* (Agilent, 2011); data reduction: *CrysAlis PRO* (Agilent, 2011); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

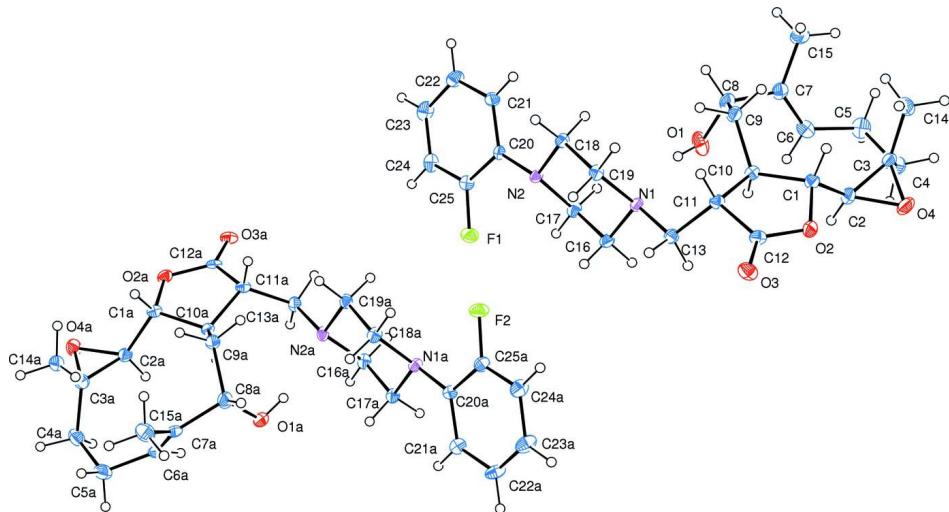
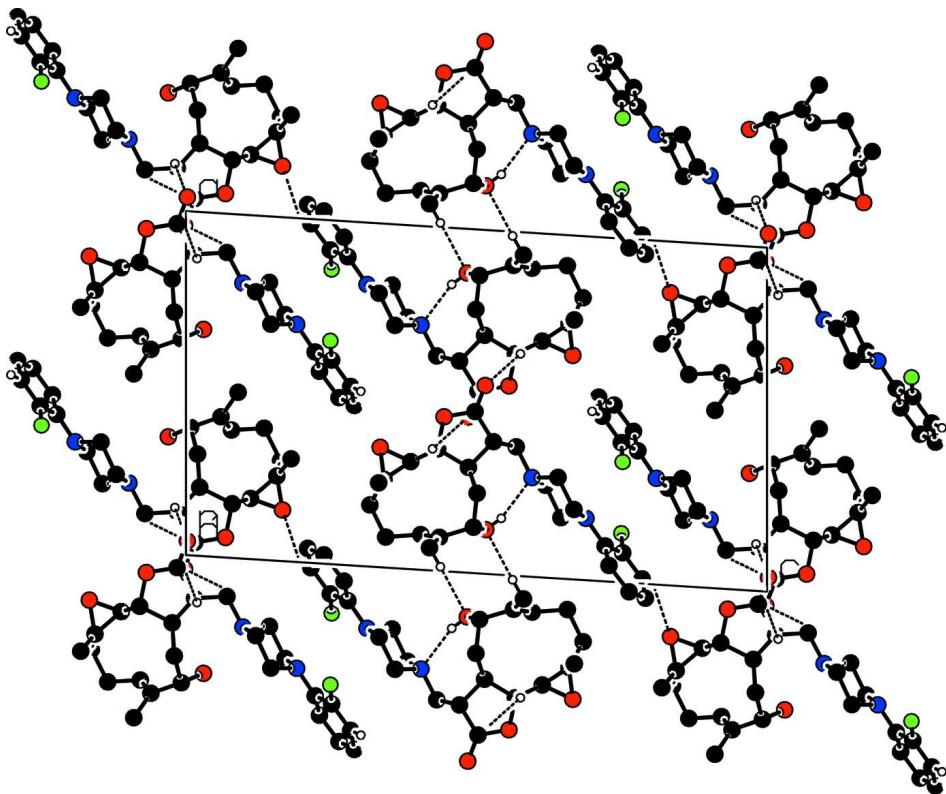


Figure 1

Molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level. H atoms are represented as small spheres of arbitrary radii.

**Figure 2**

Partial packing view showing the C–H \cdots O hydrogen bonds as dashed lines. H atoms not involved in hydrogen bonding have been omitted for clarity.

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Crystal data

C₂₅H₃₃FN₂O₄

M_r = 444.53

Monoclinic, P2₁

Hall symbol: P 2yb

a = 14.583 (2) Å

b = 6.3366 (17) Å

c = 24.697 (3) Å

β = 93.598 (14) $^\circ$

V = 2277.7 (8) Å³

Z = 4

Data collection

Agilent Xcalibur Eos Gemini ultra diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 16.1978 pixels mm⁻¹

ω scans

F(000) = 952

D_x = 1.296 Mg m⁻³

Mo K α radiation, λ = 0.71073 Å

Cell parameters from 24632 reflections

θ = 3.5–26.4 $^\circ$

μ = 0.09 mm⁻¹

T = 180 K

Block, colourless

0.56 × 0.36 × 0.1 mm

Absorption correction: multi-scan
(*CrysAlis PRO*; Agilent, 2011)

T_{min} = 0.789, T_{max} = 1.000

24656 measured reflections

5052 independent reflections

4582 reflections with $I > 2\sigma(I)$

R_{int} = 0.066

$\theta_{\max} = 26.4^\circ$, $\theta_{\min} = 3.5^\circ$
 $h = -18 \rightarrow 18$

$k = -7 \rightarrow 7$
 $l = -30 \rightarrow 30$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.068$

$wR(F^2) = 0.197$

$S = 1.12$

5050 reflections

583 parameters

1 restraint

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0994P)^2 + 2.617P]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.002$

$\Delta\rho_{\max} = 0.43 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.37 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm. CrysAlisPro (Agilent Technologies, 2011)

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1A	0.8424 (3)	-0.1376 (8)	-0.07964 (18)	0.0227 (10)
H1A	0.8190	-0.0141	-0.1000	0.027*
C2A	0.8236 (3)	-0.3358 (8)	-0.11130 (18)	0.0250 (10)
H2A	0.8336	-0.4649	-0.0899	0.030*
C3A	0.7535 (3)	-0.3580 (9)	-0.15656 (19)	0.0270 (11)
C4A	0.7080 (4)	-0.5696 (10)	-0.1628 (2)	0.0353 (13)
H4A1	0.7490	-0.6773	-0.1473	0.042*
H4A2	0.6962	-0.6005	-0.2011	0.042*
C5A	0.6168 (4)	-0.5769 (10)	-0.1346 (2)	0.0369 (13)
H5A1	0.5703	-0.4981	-0.1559	0.044*
H5A2	0.5962	-0.7220	-0.1324	0.044*
C6A	0.6282 (4)	-0.4847 (9)	-0.0784 (2)	0.0314 (12)
H6A	0.6657	-0.5600	-0.0535	0.038*
C7A	0.5916 (3)	-0.3105 (10)	-0.0603 (2)	0.0301 (12)
C8A	0.6277 (3)	-0.2092 (10)	-0.00794 (19)	0.0313 (12)
H8A	0.5774	-0.1310	0.0073	0.038*
C9A	0.7044 (3)	-0.0507 (8)	-0.01998 (19)	0.0232 (10)
H9A1	0.6876	0.0201	-0.0540	0.028*
H9A2	0.7080	0.0557	0.0083	0.028*
C10A	0.8004 (3)	-0.1502 (8)	-0.02369 (17)	0.0197 (9)
H10A	0.7951	-0.3001	-0.0146	0.024*
C11A	0.8758 (3)	-0.0569 (8)	0.01478 (18)	0.0212 (9)

H11A	0.8639	0.0942	0.0191	0.025*
C12A	0.9618 (3)	-0.0835 (7)	-0.01515 (19)	0.0210 (10)
C13A	0.8871 (3)	-0.1570 (9)	0.07060 (17)	0.0250 (10)
H13A	0.9092	-0.3004	0.0671	0.030*
H13B	0.9330	-0.0790	0.0926	0.030*
C14A	0.7011 (4)	-0.1738 (9)	-0.1810 (2)	0.0303 (11)
H14A	0.7399	-0.0512	-0.1798	0.045*
H14B	0.6482	-0.1471	-0.1608	0.045*
H14C	0.6818	-0.2052	-0.2180	0.045*
C15A	0.5173 (4)	-0.1808 (11)	-0.0897 (2)	0.0382 (13)
H15A	0.5068	-0.2325	-0.1261	0.057*
H15B	0.5363	-0.0358	-0.0907	0.057*
H15C	0.4617	-0.1915	-0.0711	0.057*
C16A	0.8080 (3)	-0.3108 (8)	0.14370 (18)	0.0235 (10)
H16A	0.8558	-0.2650	0.1702	0.028*
H16B	0.8249	-0.4487	0.1305	0.028*
C17A	0.7173 (3)	-0.3263 (8)	0.17053 (18)	0.0226 (10)
H17A	0.6700	-0.3781	0.1445	0.027*
H17B	0.7232	-0.4254	0.2005	0.027*
C18A	0.6868 (3)	0.0355 (8)	0.14613 (19)	0.0244 (10)
H18A	0.6717	0.1734	0.1601	0.029*
H18B	0.6389	-0.0045	0.1191	0.029*
C19A	0.7782 (3)	0.0465 (8)	0.12006 (19)	0.0240 (10)
H19A	0.7747	0.1502	0.0911	0.029*
H19B	0.8260	0.0901	0.1468	0.029*
C20A	0.6142 (3)	-0.1175 (8)	0.22233 (18)	0.0204 (10)
C21A	0.5586 (3)	-0.2906 (9)	0.22979 (19)	0.0263 (11)
H21A	0.5701	-0.4161	0.2119	0.032*
C22A	0.4857 (3)	-0.2809 (9)	0.2634 (2)	0.0325 (12)
H22A	0.4501	-0.4005	0.2681	0.039*
C23A	0.4654 (4)	-0.0980 (10)	0.2898 (2)	0.0358 (13)
H23A	0.4156	-0.0920	0.3115	0.043*
C24A	0.5203 (4)	0.0784 (9)	0.2835 (2)	0.0320 (12)
H24A	0.5080	0.2040	0.3011	0.038*
C25A	0.5934 (3)	0.0646 (8)	0.25078 (19)	0.0251 (10)
O1A	0.6593 (3)	-0.3633 (8)	0.03028 (14)	0.0375 (10)
H1A1	0.6922	-0.3071	0.0543	0.056*
O2A	0.9417 (2)	-0.1221 (6)	-0.06803 (13)	0.0256 (7)
O3A	1.0412 (2)	-0.0722 (6)	0.00239 (14)	0.0281 (8)
O4A	0.8513 (2)	-0.3533 (7)	-0.16622 (13)	0.0319 (8)
N1A	0.6909 (3)	-0.1189 (7)	0.19052 (15)	0.0215 (8)
N2A	0.8010 (3)	-0.1602 (7)	0.09816 (14)	0.0218 (8)
F2	0.6494 (2)	0.2337 (5)	0.24852 (12)	0.0347 (7)
C1	0.6454 (3)	0.8975 (9)	0.56683 (18)	0.0250 (10)
H1	0.6460	1.0484	0.5755	0.030*
C2	0.6873 (3)	0.7724 (9)	0.61320 (18)	0.0291 (11)
H2	0.7046	0.6286	0.6034	0.035*
C3	0.7433 (3)	0.8598 (10)	0.65909 (19)	0.0272 (11)
C4	0.8206 (4)	0.7248 (10)	0.6835 (2)	0.0347 (13)

H4A	0.8043	0.5771	0.6792	0.042*
H4B	0.8293	0.7541	0.7220	0.042*
C5	0.9112 (4)	0.7675 (11)	0.6563 (2)	0.0369 (13)
H5A	0.9366	0.9009	0.6693	0.044*
H5B	0.9551	0.6574	0.6666	0.044*
C6	0.8975 (3)	0.7746 (9)	0.5959 (2)	0.0312 (12)
H6	0.8841	0.6467	0.5787	0.037*
C7	0.9021 (3)	0.9422 (10)	0.5638 (2)	0.0296 (11)
C8	0.8663 (3)	0.9337 (10)	0.5053 (2)	0.0306 (12)
H8	0.9049	1.0253	0.4843	0.037*
C9	0.7670 (3)	1.0168 (9)	0.49891 (19)	0.0269 (11)
H9A	0.7541	1.0603	0.4616	0.032*
H9B	0.7622	1.1408	0.5216	0.032*
C10	0.6936 (3)	0.8562 (8)	0.51388 (17)	0.0212 (9)
H10	0.7232	0.7176	0.5172	0.025*
C11	0.6141 (3)	0.8380 (8)	0.47070 (18)	0.0220 (10)
H11	0.6087	0.9701	0.4501	0.026*
C12	0.5300 (3)	0.8096 (8)	0.50258 (19)	0.0231 (10)
C13	0.6249 (3)	0.6545 (9)	0.43165 (18)	0.0254 (10)
H131	0.6225	0.5228	0.4515	0.030*
H132	0.5739	0.6556	0.4045	0.030*
C14	0.7548 (4)	1.0945 (10)	0.6683 (2)	0.0343 (13)
H14D	0.7033	1.1678	0.6511	0.051*
H14E	0.8103	1.1413	0.6533	0.051*
H14F	0.7579	1.1232	0.7066	0.051*
C15	0.9342 (4)	1.1565 (11)	0.5808 (2)	0.0395 (14)
H15D	0.9490	1.1575	0.6192	0.059*
H15E	0.8864	1.2573	0.5721	0.059*
H15F	0.9877	1.1928	0.5621	0.059*
C16	0.7315 (4)	0.4575 (8)	0.3813 (2)	0.0272 (11)
H16C	0.6828	0.4194	0.3545	0.033*
H16D	0.7337	0.3517	0.4097	0.033*
C17	0.8227 (3)	0.4615 (8)	0.35471 (19)	0.0261 (11)
H17C	0.8719	0.4938	0.3817	0.031*
H17D	0.8347	0.3238	0.3395	0.031*
C18	0.7965 (3)	0.8276 (8)	0.33228 (18)	0.0219 (10)
H18C	0.7911	0.9279	0.3026	0.026*
H18D	0.8448	0.8760	0.3581	0.026*
C19	0.7061 (3)	0.8165 (8)	0.35974 (18)	0.0229 (10)
H19C	0.6913	0.9548	0.3736	0.027*
H19D	0.6573	0.7753	0.3334	0.027*
C20	0.8968 (3)	0.6153 (8)	0.27872 (17)	0.0192 (9)
C21	0.9531 (3)	0.7883 (9)	0.27203 (18)	0.0250 (10)
H21	0.9426	0.9125	0.2907	0.030*
C22	1.0259 (3)	0.7785 (11)	0.2375 (2)	0.0321 (12)
H22	1.0636	0.8953	0.2337	0.039*
C23	1.0416 (3)	0.5970 (11)	0.2094 (2)	0.0343 (13)
H23	1.0898	0.5908	0.1866	0.041*
C24	0.9860 (3)	0.4236 (10)	0.2149 (2)	0.0289 (11)

H24	0.9969	0.2990	0.1965	0.035*
C25	0.9135 (3)	0.4382 (8)	0.24841 (19)	0.0226 (10)
O1	0.8713 (3)	0.7258 (8)	0.48416 (15)	0.0429 (11)
H1B	0.8355	0.7144	0.4574	0.064*
O2	0.5509 (2)	0.8223 (6)	0.55579 (13)	0.0275 (8)
O3	0.4522 (2)	0.7769 (7)	0.48496 (15)	0.0331 (9)
O4	0.6512 (3)	0.7882 (8)	0.66572 (13)	0.0384 (10)
N1	0.7116 (3)	0.6642 (7)	0.40444 (14)	0.0206 (8)
N2	0.8201 (3)	0.6209 (7)	0.31174 (15)	0.0197 (8)
F1	0.8570 (2)	0.2681 (5)	0.25078 (12)	0.0337 (7)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1A	0.021 (2)	0.021 (2)	0.027 (2)	-0.0018 (19)	0.0038 (17)	0.0016 (19)
C2A	0.031 (2)	0.020 (2)	0.024 (2)	-0.001 (2)	0.0039 (18)	0.000 (2)
C3A	0.031 (3)	0.025 (3)	0.025 (2)	-0.001 (2)	0.0054 (19)	-0.003 (2)
C4A	0.045 (3)	0.030 (3)	0.031 (3)	-0.001 (3)	-0.005 (2)	-0.007 (2)
C5A	0.044 (3)	0.027 (3)	0.038 (3)	-0.008 (3)	-0.005 (2)	-0.001 (2)
C6A	0.028 (3)	0.031 (3)	0.035 (3)	-0.012 (2)	-0.001 (2)	0.007 (2)
C7A	0.023 (2)	0.038 (3)	0.030 (2)	-0.017 (2)	0.0050 (18)	0.002 (2)
C8A	0.026 (2)	0.043 (3)	0.025 (2)	-0.005 (3)	0.0047 (18)	-0.001 (2)
C9A	0.019 (2)	0.026 (3)	0.025 (2)	0.000 (2)	0.0019 (16)	0.000 (2)
C10A	0.020 (2)	0.019 (2)	0.021 (2)	-0.0030 (19)	0.0028 (16)	0.0021 (18)
C11A	0.017 (2)	0.019 (2)	0.028 (2)	-0.0022 (19)	0.0030 (17)	0.0000 (19)
C12A	0.019 (2)	0.012 (2)	0.033 (2)	-0.0045 (18)	0.0051 (18)	0.0004 (19)
C13A	0.026 (2)	0.029 (3)	0.020 (2)	0.001 (2)	0.0014 (17)	-0.001 (2)
C14A	0.038 (3)	0.028 (3)	0.025 (2)	-0.004 (2)	-0.0017 (19)	0.002 (2)
C15A	0.030 (3)	0.043 (3)	0.041 (3)	-0.001 (3)	-0.006 (2)	-0.002 (3)
C16A	0.026 (2)	0.021 (2)	0.023 (2)	0.007 (2)	0.0014 (17)	0.0021 (19)
C17A	0.029 (2)	0.017 (2)	0.022 (2)	0.005 (2)	0.0024 (17)	0.0029 (19)
C18A	0.029 (2)	0.019 (2)	0.026 (2)	0.007 (2)	0.0045 (18)	0.004 (2)
C19A	0.034 (3)	0.019 (2)	0.019 (2)	0.001 (2)	0.0044 (18)	-0.0010 (18)
C20A	0.018 (2)	0.019 (2)	0.024 (2)	0.0019 (19)	-0.0004 (16)	0.0030 (19)
C21A	0.023 (2)	0.025 (3)	0.030 (2)	0.005 (2)	0.0014 (18)	0.004 (2)
C22A	0.021 (2)	0.030 (3)	0.048 (3)	-0.004 (2)	0.009 (2)	0.007 (3)
C23A	0.022 (2)	0.037 (3)	0.050 (3)	0.004 (2)	0.017 (2)	0.006 (3)
C24A	0.034 (3)	0.026 (3)	0.038 (3)	0.005 (2)	0.012 (2)	0.001 (2)
C25A	0.025 (2)	0.023 (3)	0.027 (2)	0.002 (2)	0.0029 (18)	0.004 (2)
O1A	0.041 (2)	0.046 (2)	0.0252 (18)	-0.021 (2)	0.0003 (15)	0.0071 (18)
O2A	0.0223 (17)	0.0273 (19)	0.0280 (16)	-0.0037 (15)	0.0084 (12)	0.0013 (14)
O3A	0.0203 (17)	0.0241 (19)	0.0400 (19)	-0.0054 (15)	0.0029 (14)	-0.0019 (16)
O4A	0.0317 (18)	0.039 (2)	0.0257 (17)	-0.0007 (18)	0.0085 (13)	-0.0078 (17)
N1A	0.024 (2)	0.017 (2)	0.0239 (19)	0.0023 (16)	0.0051 (15)	0.0017 (16)
N2A	0.030 (2)	0.018 (2)	0.0174 (17)	0.0042 (18)	0.0037 (14)	-0.0008 (16)
F2	0.0390 (17)	0.0206 (16)	0.0460 (17)	-0.0065 (14)	0.0151 (13)	-0.0069 (14)
C1	0.028 (2)	0.024 (2)	0.024 (2)	0.000 (2)	0.0035 (18)	0.0011 (19)
C2	0.037 (3)	0.029 (3)	0.022 (2)	-0.002 (2)	0.0046 (19)	0.003 (2)
C3	0.024 (2)	0.035 (3)	0.023 (2)	-0.005 (2)	0.0048 (18)	0.002 (2)
C4	0.042 (3)	0.035 (3)	0.026 (2)	-0.001 (3)	-0.004 (2)	0.008 (2)

C5	0.030 (3)	0.042 (3)	0.038 (3)	0.012 (3)	-0.005 (2)	0.006 (3)
C6	0.024 (2)	0.029 (3)	0.040 (3)	0.010 (2)	0.000 (2)	-0.007 (2)
C7	0.020 (2)	0.039 (3)	0.030 (2)	-0.002 (2)	0.0042 (18)	-0.001 (2)
C8	0.021 (2)	0.042 (3)	0.029 (3)	-0.003 (2)	0.0067 (19)	0.000 (2)
C9	0.025 (2)	0.033 (3)	0.022 (2)	-0.005 (2)	0.0001 (17)	0.003 (2)
C10	0.020 (2)	0.020 (2)	0.024 (2)	0.0021 (19)	0.0029 (16)	0.0028 (19)
C11	0.018 (2)	0.023 (2)	0.025 (2)	0.000 (2)	0.0033 (17)	-0.0005 (19)
C12	0.016 (2)	0.020 (3)	0.034 (2)	0.0014 (19)	0.0056 (18)	-0.004 (2)
C13	0.021 (2)	0.029 (3)	0.026 (2)	-0.004 (2)	0.0085 (17)	-0.004 (2)
C14	0.035 (3)	0.035 (3)	0.033 (3)	-0.001 (3)	0.004 (2)	-0.009 (2)
C15	0.032 (3)	0.047 (4)	0.039 (3)	-0.018 (3)	0.000 (2)	0.002 (3)
C16	0.036 (3)	0.019 (2)	0.028 (2)	-0.003 (2)	0.012 (2)	0.001 (2)
C17	0.035 (3)	0.019 (2)	0.025 (2)	0.005 (2)	0.0078 (19)	0.003 (2)
C18	0.024 (2)	0.019 (2)	0.023 (2)	0.003 (2)	0.0054 (17)	-0.0001 (19)
C19	0.025 (2)	0.021 (2)	0.024 (2)	0.006 (2)	0.0039 (17)	0.001 (2)
C20	0.018 (2)	0.024 (2)	0.016 (2)	0.002 (2)	0.0022 (15)	0.0005 (19)
C21	0.020 (2)	0.028 (3)	0.027 (2)	-0.004 (2)	0.0022 (17)	-0.001 (2)
C22	0.020 (2)	0.040 (3)	0.037 (3)	-0.004 (2)	0.0053 (19)	0.005 (3)
C23	0.019 (2)	0.050 (4)	0.035 (3)	0.005 (3)	0.0083 (19)	0.002 (3)
C24	0.028 (2)	0.032 (3)	0.027 (2)	0.013 (2)	0.0036 (18)	0.000 (2)
C25	0.019 (2)	0.022 (2)	0.026 (2)	0.004 (2)	0.0001 (17)	0.001 (2)
O1	0.032 (2)	0.060 (3)	0.036 (2)	0.014 (2)	0.0034 (16)	-0.018 (2)
O2	0.0257 (17)	0.033 (2)	0.0243 (16)	-0.0030 (16)	0.0072 (12)	-0.0023 (15)
O3	0.0195 (17)	0.040 (2)	0.040 (2)	-0.0038 (17)	0.0028 (14)	-0.0114 (18)
O4	0.0328 (19)	0.055 (3)	0.0280 (17)	-0.018 (2)	0.0059 (14)	0.0071 (19)
N1	0.0229 (19)	0.020 (2)	0.0190 (18)	-0.0011 (17)	0.0058 (14)	0.0015 (16)
N2	0.0219 (19)	0.0167 (19)	0.0212 (18)	0.0000 (17)	0.0076 (14)	-0.0002 (16)
F1	0.0417 (17)	0.0220 (16)	0.0382 (16)	-0.0036 (15)	0.0087 (13)	-0.0061 (13)

Geometric parameters (\AA , $\text{^{\circ}}$)

C1A—O2A	1.462 (5)	C1—O2	1.468 (6)
C1A—C2A	1.496 (7)	C1—C2	1.492 (7)
C1A—C10A	1.548 (6)	C1—C10	1.545 (6)
C1A—H1A	0.9800	C1—H1	0.9800
C2A—O4A	1.444 (5)	C2—O4	1.434 (6)
C2A—C3A	1.473 (7)	C2—C3	1.464 (7)
C2A—H2A	0.9800	C2—H2	0.9800
C3A—O4A	1.461 (6)	C3—O4	1.437 (6)
C3A—C4A	1.500 (8)	C3—C4	1.510 (7)
C3A—C14A	1.501 (8)	C3—C14	1.512 (8)
C4A—C5A	1.540 (8)	C4—C5	1.543 (8)
C4A—H4A1	0.9700	C4—H4A	0.9700
C4A—H4A2	0.9700	C4—H4B	0.9700
C5A—C6A	1.504 (7)	C5—C6	1.494 (7)
C5A—H5A1	0.9700	C5—H5A	0.9700
C5A—H5A2	0.9700	C5—H5B	0.9700
C6A—C7A	1.317 (8)	C6—C7	1.330 (8)
C6A—H6A	0.9300	C6—H6	0.9300
C7A—C8A	1.508 (7)	C7—C15	1.488 (9)

C7A—C15A	1.509 (8)	C7—C8	1.506 (7)
C8A—O1A	1.415 (7)	C8—O1	1.420 (8)
C8A—C9A	1.546 (7)	C8—C9	1.540 (7)
C8A—H8A	0.9800	C8—H8	0.9800
C9A—C10A	1.543 (6)	C9—C10	1.540 (7)
C9A—H9A1	0.9700	C9—H9A	0.9700
C9A—H9A2	0.9700	C9—H9B	0.9700
C10A—C11A	1.526 (6)	C10—C11	1.529 (6)
C10A—H10A	0.9800	C10—H10	0.9800
C11A—C12A	1.506 (6)	C11—C12	1.510 (6)
C11A—C13A	1.517 (6)	C11—C13	1.525 (7)
C11A—H11A	0.9800	C11—H11	0.9800
C12A—O3A	1.212 (5)	C12—O3	1.207 (5)
C12A—O2A	1.343 (6)	C12—O2	1.333 (6)
C13A—N2A	1.466 (6)	C13—N1	1.470 (5)
C13A—H13A	0.9700	C13—H131	0.9700
C13A—H13B	0.9700	C13—H132	0.9700
C14A—H14A	0.9600	C14—H14D	0.9600
C14A—H14B	0.9600	C14—H14E	0.9600
C14A—H14C	0.9600	C14—H14F	0.9600
C15A—H15A	0.9600	C15—H15D	0.9600
C15A—H15B	0.9600	C15—H15E	0.9600
C15A—H15C	0.9600	C15—H15F	0.9600
C16A—N2A	1.474 (6)	C16—N1	1.465 (7)
C16A—C17A	1.520 (6)	C16—C17	1.520 (6)
C16A—H16A	0.9700	C16—H16C	0.9700
C16A—H16B	0.9700	C16—H16D	0.9700
C17A—N1A	1.464 (6)	C17—N2	1.464 (6)
C17A—H17A	0.9700	C17—H17C	0.9700
C17A—H17B	0.9700	C17—H17D	0.9700
C18A—N1A	1.468 (6)	C18—N2	1.454 (6)
C18A—C19A	1.518 (6)	C18—C19	1.522 (6)
C18A—H18A	0.9700	C18—H18C	0.9700
C18A—H18B	0.9700	C18—H18D	0.9700
C19A—N2A	1.463 (6)	C19—N1	1.465 (6)
C19A—H19A	0.9700	C19—H19C	0.9700
C19A—H19B	0.9700	C19—H19D	0.9700
C20A—C21A	1.384 (7)	C20—C25	1.379 (7)
C20A—C25A	1.394 (7)	C20—C21	1.386 (7)
C20A—N1A	1.407 (6)	C20—N2	1.426 (5)
C21A—C22A	1.391 (7)	C21—C22	1.404 (6)
C21A—H21A	0.9300	C21—H21	0.9300
C22A—C23A	1.370 (9)	C22—C23	1.370 (9)
C22A—H22A	0.9300	C22—H22	0.9300
C23A—C24A	1.389 (8)	C23—C24	1.377 (9)
C23A—H23A	0.9300	C23—H23	0.9300
C24A—C25A	1.380 (7)	C24—C25	1.385 (6)
C24A—H24A	0.9300	C24—H24	0.9300
C25A—F2	1.351 (6)	C25—F1	1.361 (6)

O1A—H1A1	0.8200	O1—H1B	0.8200
O2A—C1A—C2A	107.8 (4)	O2—C1—C2	107.7 (4)
O2A—C1A—C10A	105.7 (3)	O2—C1—C10	105.0 (4)
C2A—C1A—C10A	110.7 (4)	C2—C1—C10	112.0 (4)
O2A—C1A—H1A	110.8	O2—C1—H1	110.6
C2A—C1A—H1A	110.8	C2—C1—H1	110.6
C10A—C1A—H1A	110.8	C10—C1—H1	110.6
O4A—C2A—C3A	60.1 (3)	O4—C2—C3	59.5 (3)
O4A—C2A—C1A	120.2 (4)	O4—C2—C1	120.1 (5)
C3A—C2A—C1A	124.9 (4)	C3—C2—C1	125.1 (5)
O4A—C2A—H2A	113.7	O4—C2—H2	113.8
C3A—C2A—H2A	113.7	C3—C2—H2	113.8
C1A—C2A—H2A	113.7	C1—C2—H2	113.8
O4A—C3A—C2A	58.9 (3)	O4—C3—C2	59.2 (3)
O4A—C3A—C4A	115.6 (5)	O4—C3—C4	117.1 (5)
C2A—C3A—C4A	116.4 (5)	C2—C3—C4	117.5 (5)
O4A—C3A—C14A	113.3 (4)	O4—C3—C14	112.9 (5)
C2A—C3A—C14A	122.8 (5)	C2—C3—C14	122.7 (5)
C4A—C3A—C14A	116.3 (4)	C4—C3—C14	115.1 (5)
C3A—C4A—C5A	111.7 (5)	C3—C4—C5	111.4 (4)
C3A—C4A—H4A1	109.3	C3—C4—H4A	109.4
C5A—C4A—H4A1	109.3	C5—C4—H4A	109.4
C3A—C4A—H4A2	109.3	C3—C4—H4B	109.4
C5A—C4A—H4A2	109.3	C5—C4—H4B	109.4
H4A1—C4A—H4A2	107.9	H4A—C4—H4B	108.0
C6A—C5A—C4A	110.9 (4)	C6—C5—C4	112.1 (4)
C6A—C5A—H5A1	109.5	C6—C5—H5A	109.2
C4A—C5A—H5A1	109.5	C4—C5—H5A	109.2
C6A—C5A—H5A2	109.5	C6—C5—H5B	109.2
C4A—C5A—H5A2	109.5	C4—C5—H5B	109.2
H5A1—C5A—H5A2	108.0	H5A—C5—H5B	107.9
C7A—C6A—C5A	128.0 (5)	C7—C6—C5	127.7 (6)
C7A—C6A—H6A	116.0	C7—C6—H6	116.2
C5A—C6A—H6A	116.0	C5—C6—H6	116.2
C6A—C7A—C8A	121.4 (5)	C6—C7—C15	125.9 (5)
C6A—C7A—C15A	125.9 (5)	C6—C7—C8	121.0 (5)
C8A—C7A—C15A	112.5 (5)	C15—C7—C8	112.8 (5)
O1A—C8A—C7A	111.1 (5)	O1—C8—C7	111.3 (5)
O1A—C8A—C9A	111.6 (4)	O1—C8—C9	110.4 (4)
C7A—C8A—C9A	109.3 (4)	C7—C8—C9	110.7 (4)
O1A—C8A—H8A	108.3	O1—C8—H8	108.1
C7A—C8A—H8A	108.3	C7—C8—H8	108.1
C9A—C8A—H8A	108.3	C9—C8—H8	108.1
C10A—C9A—C8A	114.5 (4)	C8—C9—C10	114.4 (5)
C10A—C9A—H9A1	108.6	C8—C9—H9A	108.7
C8A—C9A—H9A1	108.6	C10—C9—H9A	108.7
C10A—C9A—H9A2	108.6	C8—C9—H9B	108.7
C8A—C9A—H9A2	108.6	C10—C9—H9B	108.7

H9A1—C9A—H9A2	107.6	H9A—C9—H9B	107.6
C11A—C10A—C9A	115.2 (4)	C11—C10—C9	113.0 (4)
C11A—C10A—C1A	103.1 (3)	C11—C10—C1	103.8 (4)
C9A—C10A—C1A	116.3 (4)	C9—C10—C1	116.8 (4)
C11A—C10A—H10A	107.3	C11—C10—H10	107.6
C9A—C10A—H10A	107.3	C9—C10—H10	107.6
C1A—C10A—H10A	107.3	C1—C10—H10	107.6
C12A—C11A—C13A	110.7 (4)	C12—C11—C13	111.0 (4)
C12A—C11A—C10A	103.9 (4)	C12—C11—C10	104.5 (3)
C13A—C11A—C10A	115.9 (4)	C13—C11—C10	113.3 (4)
C12A—C11A—H11A	108.7	C12—C11—H11	109.3
C13A—C11A—H11A	108.7	C13—C11—H11	109.3
C10A—C11A—H11A	108.7	C10—C11—H11	109.3
O3A—C12A—O2A	120.3 (4)	O3—C12—O2	121.2 (4)
O3A—C12A—C11A	128.6 (4)	O3—C12—C11	127.5 (4)
O2A—C12A—C11A	111.1 (4)	O2—C12—C11	111.3 (4)
N2A—C13A—C11A	112.3 (4)	N1—C13—C11	112.5 (4)
N2A—C13A—H13A	109.1	N1—C13—H131	109.1
C11A—C13A—H13A	109.1	C11—C13—H131	109.1
N2A—C13A—H13B	109.1	N1—C13—H132	109.1
C11A—C13A—H13B	109.1	C11—C13—H132	109.1
H13A—C13A—H13B	107.9	H131—C13—H132	107.8
C3A—C14A—H14A	109.5	C3—C14—H14D	109.5
C3A—C14A—H14B	109.5	C3—C14—H14E	109.5
H14A—C14A—H14B	109.5	H14D—C14—H14E	109.5
C3A—C14A—H14C	109.5	C3—C14—H14F	109.5
H14A—C14A—H14C	109.5	H14D—C14—H14F	109.5
H14B—C14A—H14C	109.5	H14E—C14—H14F	109.5
C7A—C15A—H15A	109.5	C7—C15—H15D	109.5
C7A—C15A—H15B	109.5	C7—C15—H15E	109.5
H15A—C15A—H15B	109.5	H15D—C15—H15E	109.5
C7A—C15A—H15C	109.5	C7—C15—H15F	109.5
H15A—C15A—H15C	109.5	H15D—C15—H15F	109.5
H15B—C15A—H15C	109.5	H15E—C15—H15F	109.5
N2A—C16A—C17A	110.7 (4)	N1—C16—C17	110.7 (4)
N2A—C16A—H16A	109.5	N1—C16—H16C	109.5
C17A—C16A—H16A	109.5	C17—C16—H16C	109.5
N2A—C16A—H16B	109.5	N1—C16—H16D	109.5
C17A—C16A—H16B	109.5	C17—C16—H16D	109.5
H16A—C16A—H16B	108.1	H16C—C16—H16D	108.1
N1A—C17A—C16A	110.1 (4)	N2—C17—C16	109.9 (4)
N1A—C17A—H17A	109.6	N2—C17—H17C	109.7
C16A—C17A—H17A	109.6	C16—C17—H17C	109.7
N1A—C17A—H17B	109.6	N2—C17—H17D	109.7
C16A—C17A—H17B	109.6	C16—C17—H17D	109.7
H17A—C17A—H17B	108.2	H17C—C17—H17D	108.2
N1A—C18A—C19A	110.6 (4)	N2—C18—C19	110.1 (4)
N1A—C18A—H18A	109.5	N2—C18—H18C	109.6
C19A—C18A—H18A	109.5	C19—C18—H18C	109.6

N1A—C18A—H18B	109.5	N2—C18—H18D	109.6
C19A—C18A—H18B	109.5	C19—C18—H18D	109.6
H18A—C18A—H18B	108.1	H18C—C18—H18D	108.2
N2A—C19A—C18A	109.9 (4)	N1—C19—C18	111.0 (4)
N2A—C19A—H19A	109.7	N1—C19—H19C	109.4
C18A—C19A—H19A	109.7	C18—C19—H19C	109.4
N2A—C19A—H19B	109.7	N1—C19—H19D	109.4
C18A—C19A—H19B	109.7	C18—C19—H19D	109.4
H19A—C19A—H19B	108.2	H19C—C19—H19D	108.0
C21A—C20A—C25A	116.1 (4)	C25—C20—C21	116.9 (4)
C21A—C20A—N1A	124.3 (4)	C25—C20—N2	120.2 (4)
C25A—C20A—N1A	119.5 (4)	C21—C20—N2	122.8 (4)
C20A—C21A—C22A	121.4 (5)	C20—C21—C22	120.9 (5)
C20A—C21A—H21A	119.3	C20—C21—H21	119.5
C22A—C21A—H21A	119.3	C22—C21—H21	119.5
C23A—C22A—C21A	121.1 (5)	C23—C22—C21	120.1 (5)
C23A—C22A—H22A	119.4	C23—C22—H22	120.0
C21A—C22A—H22A	119.4	C21—C22—H22	120.0
C22A—C23A—C24A	119.0 (4)	C22—C23—C24	120.1 (4)
C22A—C23A—H23A	120.5	C22—C23—H23	119.9
C24A—C23A—H23A	120.5	C24—C23—H23	119.9
C25A—C24A—C23A	119.1 (5)	C23—C24—C25	118.8 (5)
C25A—C24A—H24A	120.5	C23—C24—H24	120.6
C23A—C24A—H24A	120.5	C25—C24—H24	120.6
F2—C25A—C24A	117.7 (5)	F1—C25—C20	119.5 (4)
F2—C25A—C20A	119.0 (4)	F1—C25—C24	117.4 (5)
C24A—C25A—C20A	123.2 (5)	C20—C25—C24	123.1 (5)
C8A—O1A—H1A1	109.5	C8—O1—H1B	109.5
C12A—O2A—C1A	110.9 (3)	C12—O2—C1	110.9 (3)
C2A—O4A—C3A	61.0 (3)	C2—O4—C3	61.3 (3)
C20A—N1A—C17A	115.4 (4)	C19—N1—C16	107.3 (3)
C20A—N1A—C18A	114.6 (4)	C19—N1—C13	111.3 (4)
C17A—N1A—C18A	110.2 (3)	C16—N1—C13	109.8 (4)
C19A—N2A—C13A	112.4 (4)	C20—N2—C18	115.7 (4)
C19A—N2A—C16A	107.7 (3)	C20—N2—C17	114.3 (4)
C13A—N2A—C16A	110.0 (4)	C18—N2—C17	111.4 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

D—H···A	D—H	H···A	D···A	D—H···A
O1—H1B···N1	0.82	2.19	2.979 (6)	163
O1A—H1A1···N2A	0.82	2.08	2.882 (6)	164
C1—H1···O3 ⁱ	0.98	2.47	3.035 (7)	116
C13—H131···O3 ⁱⁱ	0.97	2.51	3.394 (7)	152
C10—H10···O1	0.98	2.36	2.861 (6)	111
C10A—H10A···O1A	0.98	2.36	2.860 (6)	111
C13A—H13A···O3A ⁱⁱⁱ	0.97	2.57	3.391 (6)	143

supplementary materials

C11A—H11A···O3A ^{iv}	0.98	2.60	3.338 (6)	132
C15—H15F···O1 ^v	0.96	2.42	3.375 (7)	171

Symmetry codes: (i) $-x+1, y+1/2, -z+1$; (ii) $-x+1, y-1/2, -z+1$; (iii) $-x+2, y-1/2, -z$; (iv) $-x+2, y+1/2, -z$; (v) $-x+2, y+1/2, -z+1$.